

S=1/2 chain in BiVO₃F: spin-dimers versus photoanodic properties

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Supporting Information

S1. Polycrystalline phase analysis

The synthesis of single phase powder sample was tried using the protocol detailed in the Experimental part. At best, starting out of stoichiometry and after sonification stages to remove impurity, BiVO₃F can be obtained as major phase (87.5 wt%) with two additional impurities: BiF₃ (4 wt%) and Bi₇F₁₁O₅ (8.5 wt%). A Rietveld refinement was performed using the Fullprof Suite [1] using the single-crystal structural model, leading to reliability factors $R_F=0.0697$, Bragg $R_{Factor}=0.0957$ and $R_{F-factor}=0.07$ for BiVO₃F.

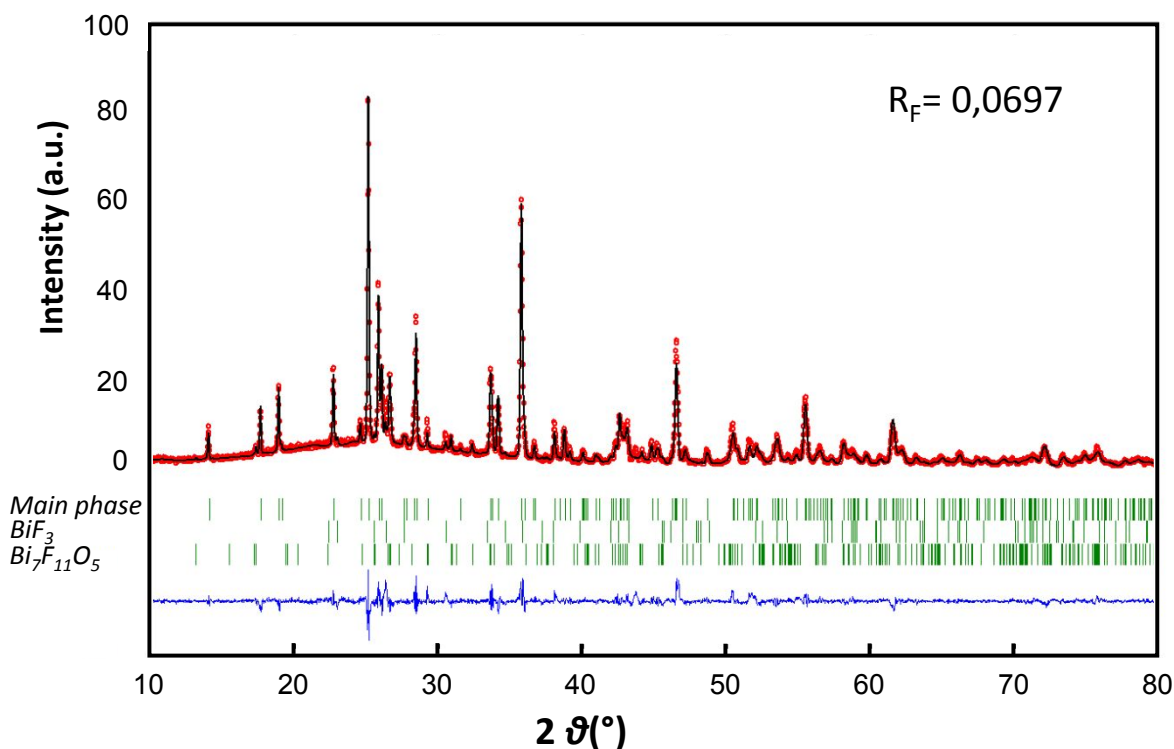


Figure S1 : XRD Rietveld refinement plots for BiVO₃F.

To date, our trials to synthesize the pure phase using solid state synthesis systematically failed. To do so, BiOF + VO₂ was weighted, mixed and sealed in tube under vacuum. It was heated at 600° for 90h and at 700° for 48h. The resulting powder sample is multiphased with evidence of AurivilliusBi₂V(O,F)₆ under investigation and hollandite Bi_xV₈O₁₆.

Table S1. Crystal and refinement data for BiVO₃F.

Crystal Data (T=293K)	
Formula	BiVO ₃ F
Molar weight (g/mol)	326.9
Symmetry	Monoclinic
Space group	P 1 2 ₁ /n 1 (14)
Unit cell (Å) and angle (°)	a = 5.2621(3) b = 4.9721(3) c = 12.6149(7) β = 95.590(2)
Volume	328.48(3)
Z	4
Data collection	
Equipment	Bruker
λ (Ag Kα (graphite monochromator); Å)	0.56087
density calc. (g/cm ³)	6.6105
Colour	black
θ (min-max) (°)	2.56 – 46.04
μ (mm ⁻¹ ; for λ Kα=0,71073Å)	30.689
R _{int} (%)	6.45
Recording reciprocal space	-13 ≤ h ≤ 13 -10 ≤ k ≤ 12 -32 ≤ l ≤ 31
Number of measured reflections	51579
Number of independent reflections (I > 3σ(I))	914/885
Crystal dimension (μm)	100*100*50
Refinement	
Number of refined parameters	56
Refinement method, program	Least square on F
Weighting scheme	Unit
R1(F)[I > 3σ(I)]/R1(F ²) [All data, %]	1.59
wR ² (F ²)[I > 3σ(I)]/wR2(F ²) [All data, %]	1.81
GOF	1,03
Max/min residual electronic density (e-/Å ³)	2.43 / -1.08
Refined extinction coefficient	5(13)

Table S2. Atomic Positions and equivalent isotropic thermal displacement for BiVO₃F.

Atom	Wick.	x	y	z	U _{eq} (Å ²)
Bi	4e	0.44052(7)	0.69997(8)	0.16736(3)	0.00830(11)
V	4e	0.5518(4)	0.7256(4)	0.43028(14)	0.0099(5)
O1	4e	0.6721(15)	0.5788(16)	0.5685(6)	0.0098(19)
O2	4e	-0.1513(15)	1.5971(16)	0.1922(6)	0.010(2)
O3	4e	-0.2265(15)	-0.3711(19)	0.3556(6)	0.014(2)
F	4e	0.7236(12)	1.0607(14)	0.4860(5)	0.0110(17)

Table S3. Anisotropic thermal displacement for BiVO₃F.

Atom	U ₁₁	U ₂₂	U ₃₃	U ₁₂	U ₁₃	U ₂₃
Bi	0.00862(19)	0.0088(2)	0.00739(19)	0.00071(12)	0.00050(11)	0.00005(12)
V	0.0135(8)	0.0070(8)	0.0081(7)	-0.0003(6)	-0.0033(6)	0.0005(6)
O1	0.016(3)	0.005(3)	0.008(3)	-0.003(3)	-0.005(3)	-0.001(3)
F	0.012(3)	0.009(3)	0.012(3)	-0.002(2)	0.003(2)	-0.002(2)
O2	0.015(3)	0.006(4)	0.008(3)	-0.005(3)	-0.001(3)	-0.001(3)
O3	0.014(4)	0.017(4)	0.012(3)	0.005(3)	0.002(3)	0.000(3)

Table S4. Bond valence sum calculations using distances < 3 Å

Atom	BVS
Bi	3.18(3)
V	4.10(3)
O1	2.26(3)
O2	2.18(3)
O3	1.87(4)
F	0.976(9)

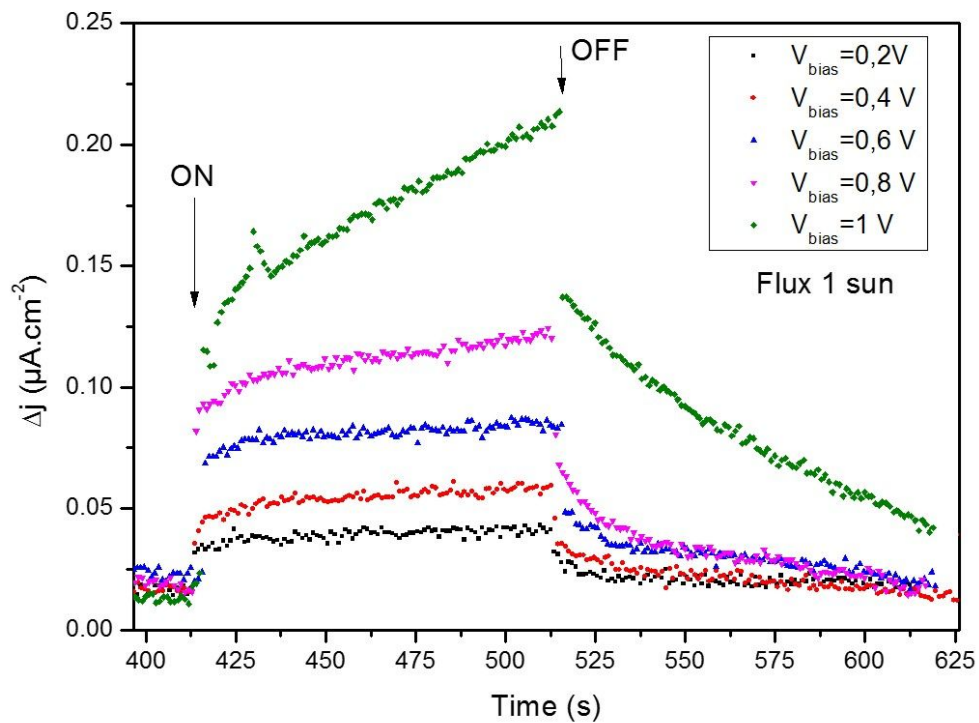


Figure S2. Variation of current density under illumination (AM1.5G one-sun ($100\text{ mW}\cdot\text{cm}^{-2}$)) or dark depending on the applied potential (V_{bias}) vs. Ag/AgCl.